Stereospecific 1,2-Aryl Migration in 2-Hydroxy Alkyl Aryl Acetals with PPh3/CCl4. Syntheses of Optically Active Ibuprofen and Naproxen*

H.R. Sonawane*, B.S. Nanjundiah, D.G. Kulkarni and J.R. Ahuja

National Chemical Laboratory, Pune 411 008, INDIA.

(Received 8 March 1991)

Abstract: Treatment of 2-hydroxy acetals (R)-(-)-1 and (R)-(-)-3 with PPh₃ and CCl₄ resulted in a stereospecific 1,2-aryl migration leading to the asymmetric synthesis of (R)-(-)-2 and (R)-(-)-4 respectively.

The use of enantiomerically pure drugs in chemotherapy is almost becoming mandatory not only to realize the enhanced specificity of drug action¹ but to avert the possible toxicity and undesirable load on the metabolism by the other enantiomer ¹. Currently, the synthesis of optically active α -arylpropanoic acids has emerged as an area of intense research² owing to their marked anti-inflammatory action. In this context, the rearrangement reactions of propiophenone derivatives are merited for their propensity for stereospecific 1,2-aryl migration and utilized in obtaining optically active acids on a practical scale³.

Generally, the reactions of alcohols with PPh₃/CCl₄ are known to offer alkyl halides without any rearrangement⁴. However, we reported recently⁵ that such a treatment of (\pm) -2-hydroxy propiophenone dimethylacetals in the presence of pyridine resulted in a rearrangement, furnishing (\pm) -methyl-2-arylpropanoates in high yields, to the total exclusion of the expected chloro-ketones (eqn. 1). We wish to demonstrate herein the potential of this reaction in realizing the synthesis of optically active Ibuprofen and Naproxen in high yields.



Equation 1

The optically active α -hydroxy acetals⁶⁻⁸ (R)-(-)-1 (82% ee) and (R)-(-)-3 (70% ee), when subjected to the reaction conditions shown in eqn. 1 furnished the corresponding optically active esters (R)-(-)-2 (82% ee) and (R)-(-)-4 (70% ee), without any significant loss of optical purity⁹, thus indicating the occurrence of stereospecific rearrangement with inversion of configuration at the carbon bearing the hydroxyl group. These results further provide an experimental support to the mechanism proposed earlier by us for the rearrangement.

In this context, the isolated report⁷ of obtaining (-)-2 by the treatment of (-)-1 with SO_2Cl_2 in the presence of base, especially at low temperatures, deserves to be mentioned. Nevertheless, the presently observed efficient and stereospecific rearrangement represents one of the very few examples of asymmetric processes with the use of PPh₃/CCl₄. Finally, this transformation provides an efficient and convenient route to optically active Ibuprofen and Naproxen.

In a typical experiment, to a premixed solution of (R)-(-)-1 (5 mmol), PPh₃ (7.5 mmol), pyridine (10 mmol) in CH₂Cl₂ (10 ml), dropwise addition of CCl₄ (25 mmol) was made at room temperature. After stirring the reaction mixture for 15 hours, the product was isolated⁵ in 78% yield. Similarly, the reaction of (R)-(-)-3 afforded (R)-(-)-4 in 81% yield.



Acknowledgements. DGK and JRA thank CSIR (India) for a Fellowship.

Reference and notes

* NCL communication number 5122.

 Konig, A. in Drug Stereochemistry, Analytical Methods and Pharmacology, Wainer, I.V.; Drayer, D.E. (Eds.), pp 113-146, Marcel Dekker, Inc. New York, 1988.

- 2. Rieu, J.P.; Boucherele, A.; Cousse, H.; Mouzin, G. Tetrahedron, 1986, 42, 4095.
- 3. Giordano, C.; Coppi, L.; Restelli, A. J. Org. Chem. 1990, 55, 5400.
- 4. Appel, R. Angew. Chem. Int. Ed. Engl. 1975, 14, 801.

5. Sonawane, H.R.; Nanjundiah, B.S.; Kulkarni, D.G.; Ahuja, J.R. Tetrahedron, 1988, 44, 7319.

6. (R)-(-)-1 and (R)-(-)-3 were prepared in five steps from (S)-ethyl lactate following known procedures

(refs. 7, 8). Higher enantiomeric purities of 1 and 3, however, could not be realized owing to known slight racemization occurring during the preparation of the α -hydroxy acetal.

- 7. Yamauchi, T.; Hattori, K.; Nakao, K.; Tamaki, K. Bull. Chem. Soc. Jpn., 1987, 60, 4015.
- 8. Schloemer, G.C. Eur. Pat. Appl. EP 81,993, cf. CA: 99: P175386b (1983).

9. The optical purity of (R)-(-)1 and (R)-(-)-3 as well as that of the products has been determined by a direct comparison of their specific rotations with those reported⁷ and by their PMR data using the \mathfrak{R} iral shift reagent, Eu(hfc)₃.